

Sulfur-Extrusive Rearrangement of α -Acylthio Ester by Lithium Amide

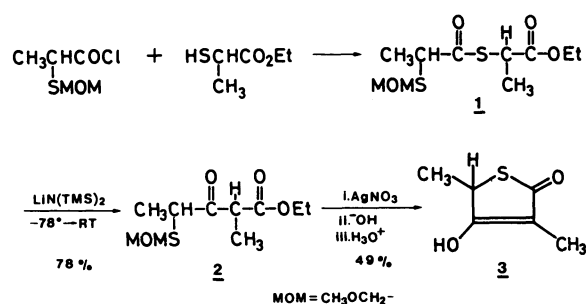
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 (Received August 11, 1984)

Synopsis. α -Acylthio esters readily undergo facile rearrangement to give β -keto esters on treatment with lithium amide at -78°C . Thiolactone **3**, a key intermediate for thiolactonic antibiotics, was synthesized by utilizing the rearrangement.

In the course of synthetic studies¹⁾ on the thiolactonic antibiotics, thiotetromycin²⁾ and thiolactomycin,³⁾ we treated ethyl 2-(benzoylthio)propanoate in dry tetrahydrofuran with 1.1 equiv lithium diisopropylamide at -78°C (20 min) to obtain the corresponding β -keto ester [m/z 206(M^+), ν (neat) 1740 and 1690 cm^{-1} , δ (CDCl_3) 1.50 (3H, d, $J=7\text{ Hz}$) and 4.39 (1H, q, $J=7\text{ Hz}$)] in 93% yield. These spectral data suggest that it is ethyl 2-benzoylpropanoate. A similar reaction, using potassium *tert*-amylate and triphenylphosphine as a desulfurating agent which removes sulfur atom of the thiirane intermediate, was reported by Eschenmoser *et al.*⁴⁾ However, in our case the additive effect of desulfurating agent was not detected completely. In order to prove the generality and limitation of this rearrangement, several S-acyl derivatives obtained by S-acylation of 2-mercapto-*propanoate* and mercaptoacetate were treated with lithium amide and the results are summarized in Table 1. α -Acylthio esters except the S-acetyl ($R^1=\text{CH}_3$) derivative undergo facile rearrangement to yield the corresponding β -keto esters without a desulfurating agent in synthetically acceptable yield.

Attention was then turned to the application of this reaction to the synthesis of a key intermediate **3** containing the latent C-2 symmetry for thiolactonic antibiotics. Compound **1**⁵⁾ obtained by acylation of 2-mercapto-*propanoate*, on exposure at -78°C with 1.1 equiv of lithium hexamethyldisilazide for 20 min

and then warming to room temperature in 1.5 h, provided β -keto ester **2**⁵⁾ as a mixture of diastereoisomers in 78% yield. Conversion of **2** to thiolactone **3**⁵⁾ was successfully achieved in 49% yield by the following sequence: (1) deprotection of methoxymethyl group with AgNO_3 ; (2) hydrolysis of ester group; (3) thiolac-



tonization by acidic work-up.

The ease of obtaining α -acylthio esters and the relatively mild conditions required for the rearrangement reaction might provide a useful tool for the construction of natural products.

References

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- 2) S. Ōmura, Y. Iwai, A. Nakagawa, R. Iwata, Y. Takahashi, H. Shimizu, and H. Tanaka, *J. Antibiot.*, **36**, 109 (1983).
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TABLE 1. SULFUR-EXTRUSIVE REARRANGEMENT OF α -ACYLTHIO ESTERS BY LITHIUM AMIDE

$\begin{array}{c} \text{O} \quad \text{H} \quad \text{O} \\ \parallel \quad \quad \parallel \\ \text{R}^1-\text{C}-\text{S}-\text{C}-\text{C}-\text{OEt} \\ \\ \text{R}^2 \end{array} \xrightarrow[\text{THF}]{\text{LiNR}_2} \begin{array}{c} \text{O} \quad \text{H} \quad \text{O} \\ \parallel \quad \quad \parallel \\ \text{R}^1-\text{C}-\text{C}-\text{C}-\text{OEt} \\ \\ \text{R}^2 \end{array}$					
Entry ^{a)}	R ¹	R ²	R ³	Reaction conditions ^{b)}	Yield/% ^{c)}
1	Ph	Me	-Pr	-78°C , 20 min	93
2	Ph	H	-Pr	-78°C RT, 1 h	76
3	$\text{Me}(\text{CH}_2)_2$	Me	Me_3Si	-78°C RT, 1.5 h	58
4	$\text{Me}(\text{CH}_2)_2$	H	Me_3Si	-78°C RT, 2 h	54
5	Me_3C	Me	Me_3Si	-78°C RT, 1.5 h	58
6	Me_3C	H	Me_3Si	-78°C RT, 2 h	59

a) Satisfactory MS, ^1H NMR and IR spectroscopic data were obtained for all compounds. b) All reactions were carried out in THF under nitrogen at -78°C for 20 min, and then warmed to room temperature. c) Isolated yield after silica-gel column chromatography.

4) M. Roth, P. Dubs, E. Gotschi, and A. Eschenmoser, *Helv. Chim. Acta*, **54**, 710 (1971).

5) **1**: MS m/z 266 (M^+); IR (neat) ν 1735, 1685 cm^{-1} ; ^1H NMR (CDCl_3) δ =1.26 (3H, t, J =7 Hz), 1.50 (6H, d, J =7 Hz), 3.33 (3H, s), 3.70 (2H, q, J =7 Hz), 4.15 (2H, q, J =7 Hz), 4.67 (2H, AB, J =12 Hz). **2**: MS m/z 234 (M^+); IR

(neat) ν 1740, 1715 cm^{-1} ; ^1H NMR (CDCl_3) δ =1.25 (3H, t, J =7 Hz), 1.34 (3H, d, J =7 Hz), 1.44 (3H, d, J =7 Hz), 3.34 (3H, s), 3.78 (1H, q, J =7 Hz), 4.13 (2H, q, J =7 Hz), 4.15 (1H, q, J =7 Hz), 4.81 (2H, s). **3**: MS m/z 144 (M^+); IR (neat) ν 1605 cm^{-1} ; ^1H NMR (CDCl_3) δ =1.63 (3H, d, J =7 Hz), 1.77 (3H, d, J =2 Hz), 4.20 (1H, qq, J =2 and 7 Hz).
